



THE IMPORTANCE OF FILM THICKNESS ON THE PHOTOVOLTAIC PERFORMANCE OF PEROVSKITE SOLAR CELLS

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ABSTRACT

In this study, we investigated the role of film thickness on the photovoltaic performance of perovskite solar cells (PSCs) fabricated from dehydrated lead acetate as the source material. The inverted p-i-n planar heterojunction architecture was adopted in this work. One step spin-coating methods were used in depositing the perovskite solution before annealing on a hot plate at 90 °C for 5 minutes to form the perovskite film. We varied the film thickness by varying the spin-coating rate from 2000 to 6000 rpm for 60 seconds and the optimal film thickness was obtained at 4000 rpm corresponding to 400 nm. This gave the best power conversion efficiency (PCE) of 11.61%, Open circuit voltage (Voc) of 0.909 V, short circuit photocurrent density (Jsc) of 20.41 mA/cm² and fill factor (FF) of 62.56%. Our investigations revealed that film thickness has great influence in the photovoltaic performance of PSCs.

Keywords: Film thickness, perovskite solar cells, spin-coating rate

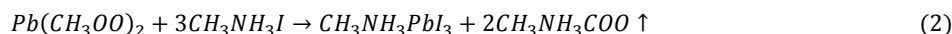
INTRODUCTION

Methylammonium lead halide perovskites are emerging as a unique group of absorbers for efficient perovskite solar cells (PSCs). This is because they display various advantages, such as inherent broad spectral absorption, ambipolar transport properties, long carrier transport diffusion length, tunable bandgap, light weight and facile solution processability (Kim et al., 2012; Lee, Teuscher, Miyasaka, Murakami, & Snaith, 2012; Stranks et al., 2013; Xing et al., 2013). Significant success has been recorded in recent years with power conversion efficiency (PCE) increased from 3.8% in 2009 to the current record of 25.2% (Feldmann et al., 2020; Kojima, Teshima, Shirai, & Miyasaka, 2009). Devices with different structures of Perovskite solar cells have been investigated and reported extensively in recent years which includes perovskite sensitized solar cells (Im, Lee, Lee, Park, & Park, 2011), mesoscopic perovskite solar cells (Q. Chen et al., 2013) and planar heterojunction perovskite solar cells (M. Liu, Johnston, & Snaith, 2013; Ye-Jin et al., 2014; Zhou et al., 2014). The planar heterojunction architecture can either be n-i-p or inverted p-i-n. Both the planar heterojunction and mesoporous architectures have recorded a high PCE (Etgar et al., 2012; M.-H. Liu et al., 2016; Tan et al., 2017; Zhou et al., 2014). The mesoporous architecture has some limitations in the choice of substrates because of the high processing temperature of over 450 °C (Burschka et al., 2013), which hinders its applications on flexible substrate and normal glass (You, Hong, Yang, Chen, Cai, Song, Chen, et al., 2014). The planar heterojunction architecture, on the other hand, is compatible with a flexible substrate (polymer) because of its low processing temperature

below 100 °C (Docampo, Ball, Darwich, Eperon, & Snaith, 2013; You, Hong, Yang, Chen, Cai, Song, & Chen, 2014). Besides the choice of the polymer substrate, the low processing temperature of the planar heterojunction architecture gives the advantages of choosing a wide range of electrodes and functional layers such as organic holes transport layers (PEDOT:PSS) and organic electron transport layers (PCBM) (You, Hong, Yang, Chen, Cai, Song, Chen, et al., 2014). Different methods have been used in depositing perovskite films such as dip-coating, spray coating, vapor deposition, vapor-assisted solution deposition, one-step spin coating process, and two-step sequential deposition (Burschka et al., 2013; Q. Chen et al., 2014; Docampo et al., 2013; Heo et al., 2013; Kim et al., 2012). The one-step spin coating and the two steps spin coating are the most widely used methods for depositing perovskite films, which are solution-processed and requires thermal or solvent annealing, hence they are cost-effective methods of producing perovskite films and these methods also allow mass production for commercial purposes.

Perovskite CH₃NH₃PbI₃ films are usually produced by the reaction of CH₃NH₃I and PbI₂ in ratio 1:1 solution as shown in equation (1). Attempts have also been made in synthesizing CH₃NH₃PbI₃ films from non-halides Pb precursors like lead acetate (Pb(OAc)₂), lead nitrate Pb(NO₃)₂, Lead (II) acetylacetonate (Pb(acac)₂) and lead (II) carbonate (PbCO₃) (Aldibaja, Fadi Kamal Badia, Laura Mas-Marzá, Elena Sánchez, Rafael S. Barea, Eva M. Mora-Sero, 2015; Andrei Buin, Patrick Pietsch, Oleksandr Voznyy & Comin, Alexander H Ip, Edward H. Sargent, 2014; Y. Chen, Yerramilli, Shen, Zhao, & Alford, 2018; Li et al., 2016; Moore, Sai, Tan, Estroff,

& Wiesner, 2014; Qing et al., 2015; Sanni et al., 2019; Sima, Vasile, & Sima, 2017; Yerramilli et al., 2018) with $Pb(OAc)_2$ been the most successful of the non-halide Pb precursor based PSCs. The equation for the reaction of lead acetate as source material is presented in equation (2) (Zhang et al., 2015). The



In this research, the one-step spin-coating method was used to investigate the role of film thickness on the photovoltaic performance of PSCs. In the spin-coating technique the thickness of the film depends on the spin coating rate (speed). We fabricated five different devices with different spin-coating rate of 2000, 3000, 4000, 5000 and 6000 rpm respectively and denoted the devices as A, B, C, D and E respectively. A power conversion efficiency (PCE) of 11.61% was obtained with a spin-coating rate of 4000 rpm.

EXPERIMENTAL

MATERIALS

Lead acetate trihydrate ($PbAc_2 \cdot 3H_2O$) was purchased from Alfa Aesar, while methylammonium iodide (MAI) was supplied by Ossila. ($PbAc_2 \cdot 3H_2O$) was processed further under a flowing nitrogen gas at 79°C to obtain the dehydrated lead acetate ($Pb(Ac)_2$). Poly (3,4-ethylene dioxythiophene): poly (styrene sulfonate) (PEDOT:PSS) was obtained from Ossila. The precursor was prepared by dissolving 3.0 mmol MAI and 1.0 mmol $Pb(Ac)_2$ in 1mL dimethylformamide (DMF) solvent.

DEVICE FABRICATION

The devices structure was ITO/PEDOT:PSS/ $CH_3NH_3PbI_3$ /PCBM/Ag. The pre-patterned glass/indium tin oxide (ITO) substrates were sequentially cleaned in a detergent solution, deionized water, acetone and isopropyl alcohol (IPA) in an ultrasonic bath for 15 minutes an hour. Finally, they were treated by exposing them to ultraviolet-ozone (UVO) for 15 minutes to get rid of any remaining impurity. The hole transport layer of PEDOT:PSS, was filtered through a 0.45 μm filter and the filtered solution was spin-coated onto the patterned ITO substrates at a spin-coating rate of 4000 rpm for 60 secs before annealing at 135 °C for 15 minutes. The solutions of MAI: $Pb(Ac)_2$ were then spin-coated on the PEDOT:PSS layer at the rate of 2000, 3000, 4000, 5000 and 6000 rpm for 60 secs (the corresponding devices are denoted as A, B, C, D and E) respectively before they were thermal anneal on the hot plate at the temperatures of 90 °C for 5 minutes. These processes resulted in perovskite films.

After that, the electron collection layer of phenyl-C₆₁-butyric acid methyl ester (PCBM, Ossila) was obtained by dissolving the solutes into a chlorobenzene solvent (20 mg/mL) and spin-coating it on $CH_3NH_3PbI_3$ layer at the rate of 1000 rpm for 30 s. Lastly, the silver (Ag) electrode was deposited on the PCBM to

complete the device fabrication process. The fabricated device is an inverted planar heterojunction solar cell (see Fig. 2(d)). The dimension of the substrate is 2.5 cm × 2.5 cm and the active area is 0.2 cm²

DEVICE CHARACTERIZATION

The current density-voltage (J-V) measurements of the solar cells devices were observed on a Keithley 2400, under simulated AM1.5G irradiation (100 mW/cm²) with a Xenon-lamp-based solar simulator (Spectra Physics, Oriel Instruments, USA)

RESULTS AND DISCUSSIONS

The device structure employed in this research to investigate the influence of perovskite film thickness on the photovoltaic properties is the inverted planar heterojunction architecture p-i-n as shown in Fig. 2(d). $Pb(Ac)_2$ is used as the Pb source with MAI to obtain the perovskite film ($CH_3NH_3PbI_3$) as stated in equation (2). The total PCE of solar cells depends on the open-circuit voltage (V_{oc}), the short circuit photocurrent density (J_{sc}), the fill factor (FF) and the intensity of the incident light (P_{in}) as represented in equation (3) (Yoshida et al., 1997). The J-V curve showing the photovoltaic performance of the perovskite solar cells derived from different perovskite film thickness is shown in Fig. 1(a-f) and (table 1) shows the device parameters.

$$PCE = \frac{J_{sc}V_{oc}FF}{P_{in}} \quad (3)$$

From the J-V curve in Fig. 1 and the summary of the photovoltaic properties in table 1 it is noted that device C fabricated with a spin-coating rate of 4000 rpm for 60 seconds has the best performing PCE of 11.61%, V_{oc} of 0.909 V, J_{sc} of 20.41 mA/cm² and FF of 62.56%, while devices A and E have the lowest PCE of 5.81% and 5.84% respectively. The reason for the low PCE of device A could be due to the poor perovskite film formation due to the low rate of deposition of 2000 rpm. At this rate, the film formation is not uniform resulting in some portion of the film to be too thick and some portion to be too thin. This poor film formation resulted in low V_{oc} and by implication low PCE as it is observed in Fig. 1(a). On the other

hand, the reason for the low PCE of device E is because at 6000 rpm the film becomes too thin which limits the amount of solar energy that can be absorbed by the photoactive layer which is evident in the low J_{sc} of device E as shown in Fig 1(e).

The main reason for the increase of PCE of device C is an increase of J_{sc} . The higher J_{sc} of a device is attributed to a smoother surface and better crystallinity of perovskite films.

The increase of J_{sc} for device C seems to be a major effect from free charge- (carrier density (n) for good quality perovskite films according to a recombination dynamic relation Wang, Chen, Han, Zhang, & Ma, 2016)

$$\frac{dn(t)}{dt} = -k_3n^3 - k_2n^2 - k_1n^1 \quad (4),$$

where k_1 , k_2 and k_3 are trap-assisted recombination rate, radiative recombination rate and Auger recombination rate, respectively. In this case, good quality films could be considered to reduce the trap-assisted recombination rate. When charge transfers through poor films, some of them might be trapped at the defect site and recombined to decrease J_{sc} . On the other hand, good films will contribute to better charge transfer with lower trapped state increasing of J_{sc} as seen in Table 1. The trapped state can affect the performance of perovskite solar cells by reducing J_{sc} due to the slow movement of electrons which inturns increase the recombination effect.

To further understand the influence of film thickness on the photovoltaic performance of PSCs, a semi-logarithmic plots of the dark J–V characteristics shown in Fig. 2(a) was employed. From the semi-log plots it is observed that devices A and E have the highest leakage current while devices C and D have the lowest leakage current. The high leakage current in devices A and E could signifies that the electrons are not well extracted from the photoactive layer. On the other hand, the decreased leakage current in devices C and D could be that the electron transport layer are effectively extracted from the perovskite layer.

The abnormal hysteresis in the J-V curves has shown to be a major challenge in evaluating perovskite solar cells (Shao, Xiao, Bi, Yuan, & Huang, 2014). In this research, we studied the hysteresis of device C to be certain of the accuracy of the measurement. We measure the forward and the reverse scan of the J-V curve of the champion cell as presented in Fig. 2(b) which shows that there is no hysteresis observe. The efficiency box plot from 15 devices shown in Fig. 2(c) indicates that the device C performed better statistically compared to devices derived from other film thickness. Devices A, B and C show better repeatability compared to devices D and E this is because there are fewer pin-holes in the perovskite films due to their relative film thickness.

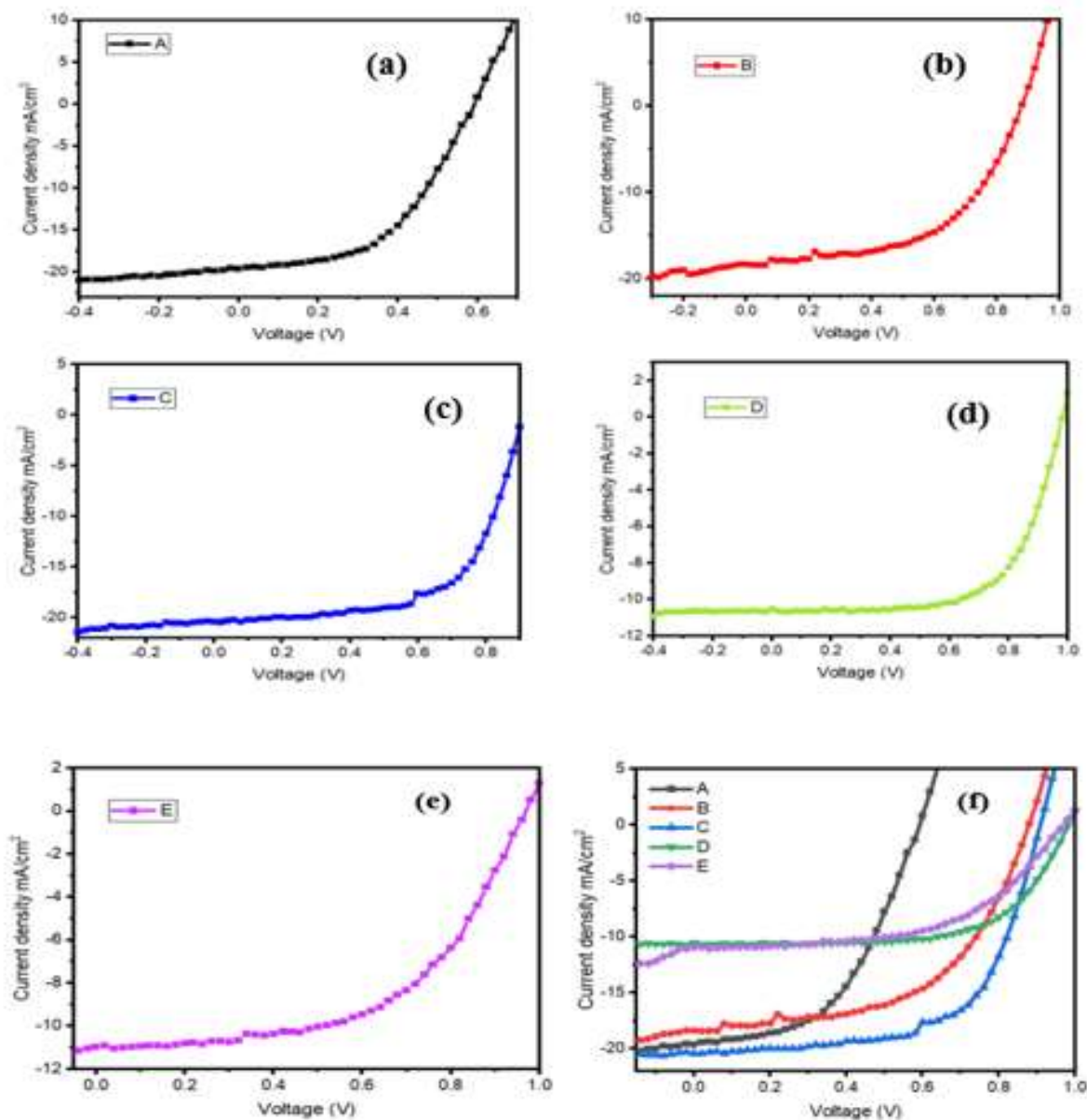


Fig. 1: J-V curves of devices derived from different spin-coating rate (a) 2000 rpm (b) 3000 rpm (c) 4000 rpm (d) 5000, (e) 6000 rpm and (f) J-V curve for the combined devices a-e

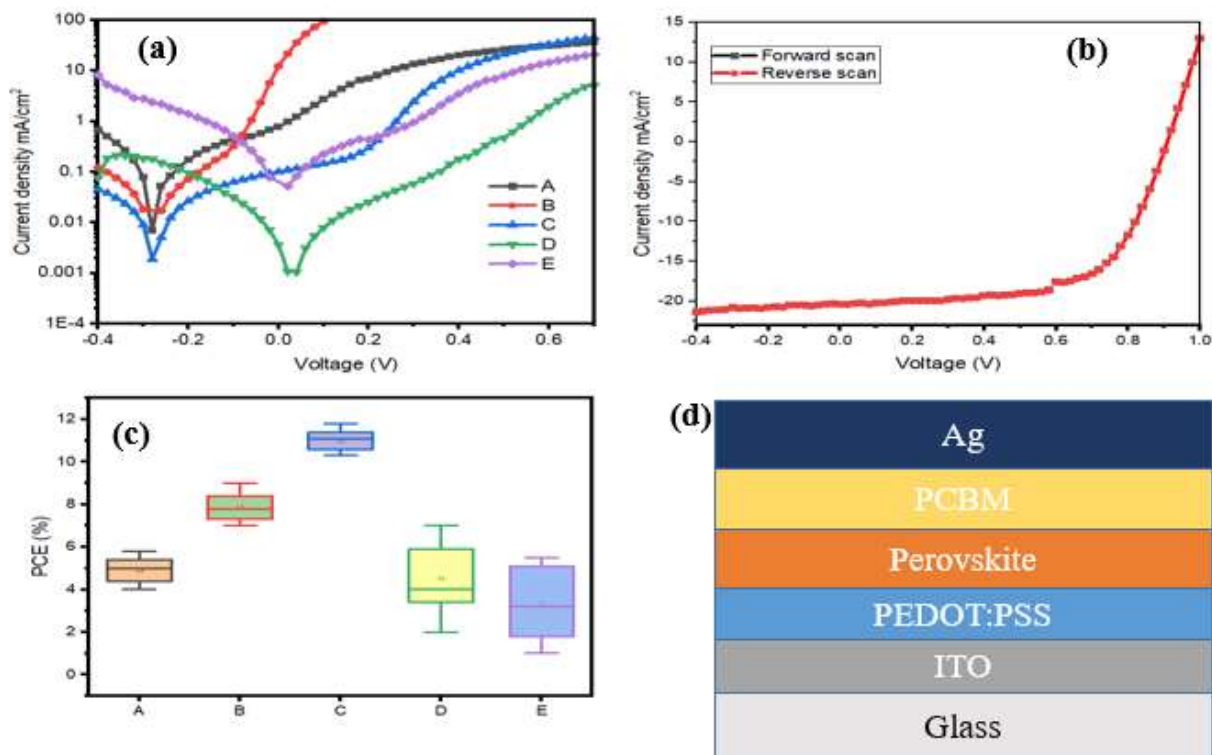


Fig. 2: (a) The semi-logarithmic plots of dark J–V characteristics for the inverted planar perovskite solar cells derived with different spin-coating rate (b) Forward and reverse J–V curves of the champion cell with a spin-coating rate of 4000 rpm (c) Box plot of 15 devices with different deposition rate (d) Schematic diagram of the inverted planar heterojunction perovskite solar cell architecture

Table 1: Photovoltaic properties for the inverted planar perovskite solar cells with different spin-coating rate

| Devices | V_{oc} (V) | J_{sc} (mA/cm ²) | FF (%) | PCE (%) | |
|---------|--------------|--------------------------------|--------|---------|--|
| A | 0.593 | 19.65 | 49.87 | 5.81 | |
| B | 0.879 | 18.38 | 54.46 | 8.80 | |
| C | 0.909 | 20.41 | 62.56 | 11.61 | |
| D | 0.982 | 10.56 | 65.63 | 6.80 | |
| E | 0.969 | 10.96 | 54.96 | 5.84 | |

CONCLUSION

We used one step spin-coating method to investigate the impact of film thickness on the photovoltaic performance of PSCs. The investigation reveals that film thickness has great effect on the performance and production of PSCs. Device C which was fabricated with a spin-coating rate of 4000 rpm for 60 seconds and a corresponding film thickness of 400 nm produce the best PCE of 11.61% while the lowest PCE is produced by device E which was fabricated with spin-coating speed of 6000 rpm and with a corresponding film thickness of 250 nm.

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