ABSTRACT
Terminalia ivorensis (TI) is used in West African ethnmedicine for the treatment of ulcers, malaria, inflammation, and wounds. Despite its widespread use, its nutritional status remains largely undetermined. This study was undertaken with the aim to evaluate the nutritional and phytochemical composition of TI. Proximate analysis of the methanol leaf extracts of TI was carried out using the AOAC methods. Elemental analysis of the methanol leaf extracts of the plant was also carried out using an Atomic Absorption Spectrometer (AAS). Identification and quantification of the bioactive components were determined by Gas Chromatography-Mass Spectrometry (GC-MS). Proximate analysis reveals that it is rich in primary nutrients such as carbohydrates (37.70±0.05%), proteins (10.94±0.03%), fats (7.97±0.12%), fibre (0.95±0.01%), ash (1.42±0.01%) and moisture content (41.02±0.21%). Mineral analysis of the extract revealed the presence of essential minerals such as calcium (641.2±0.1 mg/100g), copper (6.2±0.02 mg/100g), magnesium (186.1±0.29 mg/100g), iron (17.1±0.06 mg/100g), manganese (3.47±0.01 mg/100g), and zinc (3.8±0.01 mg/100g), which are essential for the metabolic processes and pregnancy. Forty–Seven (47) phytochemical compounds were detected by GC-MS including 1, 2, 3- Benzene triol (26.31%), Diethyl Phthalate (14.93%), D-Allose (6.96%), Anhydro-alpha-d-galactofuranose (5.09), Glycerin (6.28%) as the most predominant. The study revealed that Terminalia ivorensis could be a useful source of nutrients, minerals, and several helpful bioactive compounds that may serves as potential drug target needed for drug development.

Keywords: Terminalia ivorensis, Nutritional, Phytochemical, Gas Chromatography, Proximate

INTRODUCTION
Phytomedicine, also known as herbal medicine or botanical medicine or phytotherapy, is the use of plant-based substances for therapeutic and healing purposes (Mohd, 2019). This practice has ancient origins and has been an integral part of human healing traditions across cultures for thousands of years. The use of plants for medicinal purposes dates to prehistoric times. Archaeological evidence suggests that early humans relied on plants for food and discovered their healing properties through trial and error. Ancient civilizations, including the Sumerians, Egyptians, and Chinese, developed extensive herbal knowledge, recording their findings on clay tablets and scrolls. The earliest known herbal texts, such as the Sumerian “Pen T’ao” and the Egyptian “Ebers Papyrus,” provide insights into the plants use for healing ailments. and a variety of ailments such as malaria, asthma, diabetes, diarrhoea, fever, hypertension, and even constipation have been reported to be treated using various herbal plants, which are usually administered as poultices (a mass of soft moist herbs, applied to the body to treat sores, aches or inflammation), concoctions of various plant, infusions of teas and tinctures (a medicine made by dissolving a drug in alcohol) or as part of ingredients in porridge and soups (Doughari, 2012). Terminalia ivorensis is a large tree known as Ivory Coast almonds that is mainly found on the coasts of West Africa). Studies have shown that this plant is known to possess antioxidant, antimicrobial, antiinflammatory, wound healing, anti-psychotic, trypanocidal activities (Ben-Aza et al., 2016, Akiniyemi et al., 2006, Agbedahunsi et al., 2006, Claudine et al., 2017, Kipré et al., 2023, Annan et al., 2012). Terminalia ivorensis is reported to be useful in the treatment of ulcers, cuts, sores, wounds, general body pains, hemorrhoids, diuresis, malaria and yellow fever (Burkill, 1985, Ouattara et al., 2013). Despite its widespread use, the phytochemical profile and bioactive components of TI remain largely undetermined. Although the composition of the nutritional component varies in plants, a study has shown that plants have the right quantity to meet human nutritive requirements (Suroowan, et al., 2019). This study investigated the nutritional composition and elemental composition of the methanol leaf extract of Terminalia ivorensis.

MATERIALS AND METHODS
Collection of plant materials
The fresh leaves of Terminalia ivorensis were pruned from the trees of Terminalia ivorensis at the open field of Department of Pharmacy, University of Benin, Benin City, Edo state, Nigeria. The leaves of Terminalia ivorensis were identified by Professor Akinibosun of the Botany Department, Faculty of Life Sciences, University of Benin, Benin-city, Edo state, Nigeria, with the voucher number, UBH-T2161.

Preparation of the Extracts
Freshly collected leaves of Terminalia Ivorensis were dried under the sun for 14 days and milled with a Bench-top milling machine in the Pharmacognosy Department, Faculty of Pharmacy, University of Benin into fine powder (1100 g) and soaked in 15 L of methanol and stirred for 3 days. The methanol solution was filtered thrice using several folded clean pieces of cheesecloth during each filtration process to remove debris. The mixture was concentrated using a rotary evaporator set at 60°C and Freeze-dried to a constant weight with a freeze drier. The dried extract gave a yield of 31.14% (w/w) and was stored in an air-tight container in a refrigerator at about 4°C until when needed.
Proximate Analysis

Proximate analysis was carried out on the methanol extract of the leaves of Terminalia Ivorensis to ascertain its nutritional composition. Analysis of the sample was carried out using the method of association of official analytical chemists, (AOAC, 1990).

Ash Content

When organic matter is burnt in a furnace for three hours at a temperature of 600°C, the amount of inorganic residue left over is measured as the sample’s ash content. The sample’s ash content was determined using the AOAC (1990) method, which involved loading 2g of the sample into an empty, pre-weighed crucible (w0), which was then heated in a muffle furnace set to 600°C for three hours. The ash was then allowed to cool in a desiccator before being reweighed (w2). By comparing the weight of the sample before and after ashing, it was possible to determine the weight of the ash. Percentage ash was obtained by:

\[
\text{%Ash} = \frac{(W_1 - W_2) \times 100}{W_0}
\]

Where: \(W_0=\text{Weight of the empty crucible, g}\)
\(W_1=\text{Weight of ashed sample, g}\)
\(W_2=\text{Weight of sample + empty beaker, g}\)

Crude fibre determination

This was done following the AOAC (1990) protocol. A 1 liter conical flask (w0) was filled with 2 g of the samples. The flask was filled with 200 ml of 1.25% H2SO4 acid, which was then allowed to boil for 30 minutes while being constantly monitored by cooling fingers and then filtered using a poplin cloth. Using a spatula, the residue was transferred back to the flask, and 200 ml of 1.25% NaOH (produced by dissolving 100g of NaOH pellets in 1 liter of distilled water) was added. The mixture was then allowed to boil for an additional 30 minutes while being constantly agitated with cooling fingers. A poplin cloth was used to filter the material, and the residue was properly cleaned with hot distilled water before being thoroughly rinsed twice with industrial methylated spirits, acetone, or ethanol. Rinse three times with petroleum ether (BP 40-60°C) to finish. Heat dry overnight at 105°C in the oven, allow it to drain completely and scrape the residue into a crucible. Once removed, it was chilled in a desiccator. In a muffle furnace, the weighed sample (w1) was ashed for 90 minutes at 550°C. In the end, it was dried in a desiccator and weighted once more (w2). The percentage crude fibre was calculated thus;

\[
\text{%of crude fibre} = \frac{(W_1 - W_2) \times 100}{W_0}
\]

\(W_0=\text{Weight of sample, g}\)
\(W_1=\text{Weight of dried sample, g}\)
\(W_2=\text{weight of ash sample, g}\)

Crude protein determination

For the purpose to determining crude protein, the AOAC (1997) employed was a modified version of the micro-Kjeldahl method. A small amount of anti-bumping granules was added to a micro-Kjeldahl digestion flask along with three grams of each of the defatted samples that were individually weighed on pre-weighed scales. Each flask received two grams of the catalyst solution (CuSO4: Na2SO4: SeO2, 5:1:02 w/w), along with 10 ml of nitrogen-free concentrated H2SO4. The flasks were set up on a heating mantle in a fume cupboard at an angle. After the foaming stopped, the temperature was raised to 50°C for another 30 minutes, then to full heating (100°C), where it remained until a clear solution was achieved. In order to achieve thorough digestion and nitrogen conversion to ammonium sulphate, simmering was maintained below boiling point for an additional 30 minutes. Samples were transferred quantitatively into 100 ml volumetric flasks after washing and cooling to room temperature once digestion was finished. With distilled water, volumes were adjusted to the proper levels. A 10 ml pipette was used to transfer 10 ml of the digest filtrate into a 25 ml standard flask. Alkaline phenate (2 ml) was added, and the mixture was agitated to ensure good mixing. After properly shaking, 2ml of sodium hypochlorite was added, followed by 6ml of sodium potassium tartrate. With the use of a UV/visible spectrophotometer, the solution’s absorbance at 630 nm was measured after being diluted with distilled water to the 25 ml mark. The sample and the nitrogen standards were handled in the same manner.

\[
\text{calculation}
\]

\[
\%N = \frac{\text{Instrument Reading \times Slope Reciprocal \times Color Vol.\times Digest Vol.}}{\text{Weight of Sample \times Aliquot Taken \times 10000}}
\]

\(\% \text{ Crude Protein} = \% \text{Nitrogen}\times6.25\)

Crude Fat

AOAC (1973)’ s soxhlet extraction method was used to estimate the crude lipid. Prior. Before weighing the empty 250 ml extraction flask, dry it in the oven at 110°C, then let it cool in the desiccator. A porous thimble with a label was filled with cotton wool, and the folded filter paper (w0) containing 2 grams of the sample was placed inside. In the dry 250-ml extraction flask, add about 200 ml of petroleum ether. Construct the apparatus by inserting the cover porous thimble into the condenser. 5–6 hours should be used for the extraction. Carefully remove the porous thimble, then retrieve the petroleum ether and store it in the top container (tube) for future use. Drying takes place in ovens at 1050°F–1100°F for a period of one hour. Take the weight after cooling in the desiccator. When the petroleum ether is nearly completely gone, remove the extraction flask from the water bath. Calculation:

\[
\text{Weight of empty porous thimble} = w_0
\]

\[
\text{Weight of empty thimble + ground sample} = W_1
\]

\[
\text{Weight of ground sample} = W_1 - W_0
\]

\[
\text{Weight of empty extraction flask} = W_2
\]

\[
\text{Weight of empty extraction flask + ether} = W_3
\]

\[
\text{Weight of ether (fat or oil)} = W_3 - W_2
\]

\[
\text{%Fat} = \frac{W_3 - W_2}{W_1 - W_0} \times 100
\]

Moisture content

This represents the amount of moisture that has evaporated after drying at 105°C in an oven. A pre-weighed (w0) crucible (w1) was filled with one (1) gram of the sample. The crucible containing the weighed sample was kept in a 105°C oven for three hours. After being taken out of the oven, the crucible was placed in a desiccator where it was permitted to cool. The weight was taken after cooling. Repeating the drying, cooling, and weighing steps led to the achievement of a consistent weight (w2). The percentage of moisture was obtained by the equation:

\[
\text{%moisture} = \frac{(W_1 - W_2) \times 100}{W_0} - \text{Weight of empty crucible, g}
\]
\[
W_1 = \text{weight of sample + empty beaker, g}
\]
\[
W_2 = \text{weight of dried sample + empty beaker, g}
\]
Determination of soluble carbohydrate (nitrogen-free extractive)

Test method no.: qal/am/po6

The nitrogen free extractive (N.F.E.), also known as soluble carbohydrate, is obtained as a difference between crude protein and the total of ash, protein, crude fat, and crude fiber rather than being directly assessed. N.F.E. 100= (% Ash + % crude fibre + % crude fat + crude protein)

N.F.E. is a general term that refers to all starches and sugars, a small amount of hemicellulose, and various amounts of lignin. It does not refer to any specific compound or collection of substances.

Elemental analysis

Nitric-Perchloric acid digestion via the Atomic Absorption Spectroscopy (AAS) technique was used to analyze specific trace minerals. About 50 mg of dried methanol leaf extract of *Terminalia ivorensis* was weighed into a 250 ml conical flask with 5 ml of Nitric-Perchloric acid mixture and allowed to stand for 24 hours. After digestion of the dried extract, known elemental and blank samples were also prepared. The digested samples and blank were run on the AAS to obtain the absorbance values. Concentrations of the elements in the samples were calculated from the standard.

GC-MS Analysis

About 100 g of the dried methanol leaf extract of *Terminalia ivorensis* was dissolved in 100 mL distilled water, further diluted with water 1:1 and filtered through a 0.22 µm PVDF filter. GC-MS analysis of *Terminalia ivorensis* was performed using a Perkin-Elmer GC Clarus 500 system and Gas chromatograph interfaced to a Mass Spectrometer (GC-MS) equipped with an Elite-I, fused silica capillary column (30 mm x 0.25 mm 1D x 1 µMdf), composed of 100% Dimethylpolysiloxane) For GC-MS detection, an electron ionization system with ionizing energy in a single-phase ion mode of 70 eV was used. Helium gas (99.999%) was used as the carrier gas at a constant flow rate of 1 mL/min and an injection volume of 0.50 ml was employed (split ratio of 10:1). Injector temperature 250°C; Ion-source temperature 280°C. The oven temperature was programmed from 110°C (isothermal for 2 min.), with an increase of 10°C/min, to 200°C, then 5°C/min to 280°C, ending with a 9 min isothermal at 280°C. Mass spectra were taken at 70 eV, a scan-interval of 0.5 seconds and fragments from 45 to 450 Da. The total GC running time was 27 minutes. The relative % amount of each component was calculated by comparing its average peak area to the total areas. Software adapted to handle mass spectra and chromatograms was a Turbo mass.

### Identification of components

Interpretation of mass spectrum from GC-MS was conducted using the database of National Institute of Standard and Technology (NIST) and has more than 62,000 patterns. The spectrum of the unknown component was compared with the spectrum of the known and authentic samples stored in the NIST library. Computer searches in an HP Mass Spectral Library NIST98 were also applied. The name, molecular weight, and structure of the components of the test materials were ascertained.

### RESULTS AND DISCUSSION

The methanol extract of the leaves of *Terminalia ivorensis* was analyzed using proximate analysis for their nutritional contents, and the results obtained are presented in Table 1.0. Analysis of the samples showed that moisture content was high (41.02%), second to carbohydrates (36.70%), and the least was crude fibre (0.95%). Protein, crude fat, and ash content were also obtained.

<table>
<thead>
<tr>
<th>Parameters (units)</th>
<th>MC (%)</th>
<th>PROTEIN (%)</th>
<th>CRUDE FAT (%)</th>
<th>CRUDE FIBRE (%)</th>
<th>AC (%)</th>
<th>NFE (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Plant extract</td>
<td>41.02±0.21</td>
<td>10.94±0.03</td>
<td>7.97±0.12</td>
<td>0.95±0.01</td>
<td>1.42±0.01</td>
<td>37.70±0.05</td>
</tr>
</tbody>
</table>

Results are expressed as means ± SEM of triplicate replication.

MC- Moisture Content, AC- Ash Content, NFE- Nitrogen Free Extract

*T. ivorensis* (coastal almonds) are rich in all primary nutrients. The percentage of proteins (10.3%) is high bearing in mind that this is a leaf extract (meat and fish possess between 15 and 20 g of proteins for each 100 g), crude fat of 7.9% with less amount of carbohydrate chemically bound to other molecules in the form of NFE (37.70), MC gave the highest amount of 41 % which suggests a very poor shelf life of the leaf if not properly stored.

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Calcium (Ca)</th>
<th>Copper (Cu)</th>
<th>Magnesium (Mg)</th>
<th>Iron (Fe)</th>
<th>Manganese (Mn)</th>
<th>Zinc (Zn)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Concentrations Mg/100g</td>
<td>641.2±1.20</td>
<td>6.28±0.02</td>
<td>186.16±0.29</td>
<td>17.13±0.06</td>
<td>3.47±0.01</td>
<td>3.82±0.01</td>
</tr>
</tbody>
</table>

Results are expressed as means ± SEM of triplicate replication

Mineral analysis of *T. ivorensis* (coastal almonds) showed almonds are among the richest plant-based foods in calcium. They also contain significant amount of magnesium, potassium iron and zinc. The calcium content of *T. ivorensis* (641 mg/100) is much greater than milk, although the amounts of extract taken is usually less than that of milk or dairy products.
Each peak represents a distinct compound. The area covered by each peak correlates with the intensities of ions at specific mass to charge m/z values which corresponds to the relative abundance (concentration) of that compound. The higher the ion intensities the larger the peak area, which in turn contributes to a higher area percentage. The identified compound corresponding to each peak with its area % value is presented in Table 4. The total analysis time per injection on the GC-MS system as shown in the chromatogram was 27 minutes.
Table 3: Compounds identified in the methanol leaf extracts of *Terminalia ivorensis*

<table>
<thead>
<tr>
<th>Compound Name;</th>
<th>Retention time(min)</th>
<th>Area (%)</th>
<th>Chemical structure</th>
<th>Molecular weight(g/mol)</th>
<th>Compound type</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cyclohexanol</td>
<td>8.675</td>
<td>0.49</td>
<td><img src="image" alt="Cyclohexanol" /></td>
<td>154</td>
<td>Terpene</td>
</tr>
<tr>
<td>Citronellol</td>
<td>8.931</td>
<td>0.31</td>
<td><img src="image" alt="Citronellol" /></td>
<td>156</td>
<td>Monoterpene alcohol</td>
</tr>
<tr>
<td>2,6-Octadien-1-ol, 3,7-dimethyl-, (Z)-;</td>
<td>9.197</td>
<td>0.33</td>
<td><img src="image" alt="2,6-Octadien-1-ol" /></td>
<td>154</td>
<td>Terpene Alcohol</td>
</tr>
<tr>
<td>1,2,3-Benzentriol</td>
<td>10.365</td>
<td>26.31</td>
<td><img src="image" alt="1,2,3-Benzentriol" /></td>
<td>126</td>
<td>Phenol</td>
</tr>
<tr>
<td>Propan-2-one</td>
<td>11.245</td>
<td>0.68</td>
<td><img src="image" alt="Propan-2-one" /></td>
<td>190</td>
<td>Ketone</td>
</tr>
<tr>
<td>D-Allose</td>
<td>11.447</td>
<td>6.96</td>
<td><img src="image" alt="D-Allose" /></td>
<td>180</td>
<td>Aldose</td>
</tr>
<tr>
<td>Tetradec-11-en-1-ol acetate</td>
<td>11.642</td>
<td>0.72</td>
<td><img src="image" alt="Tetradec-11-en-1-ol acetate" /></td>
<td>268</td>
<td>Oxirane ring</td>
</tr>
<tr>
<td>3,3-Dimethyl-4-phenyl-4-penten-2-one</td>
<td>12.080</td>
<td>0.45</td>
<td><img src="image" alt="3,3-Dimethyl-4-phenyl-4-penten-2-one" /></td>
<td>188</td>
<td>Ketone</td>
</tr>
<tr>
<td>Lilial</td>
<td>12.249</td>
<td>0.90</td>
<td><img src="image" alt="Lilial" /></td>
<td>204</td>
<td>Aldehyde</td>
</tr>
<tr>
<td>Compound</td>
<td>R1</td>
<td>R2</td>
<td>R3</td>
<td>R4</td>
<td>R5</td>
</tr>
<tr>
<td>----------------------------------------------------</td>
<td>-----</td>
<td>-----</td>
<td>-----</td>
<td>-----</td>
<td>-----</td>
</tr>
<tr>
<td>Benzene, 1-(1-methyl-2-propenyl)</td>
<td>12.468</td>
<td>0.27</td>
<td>188</td>
<td>Benzene</td>
<td></td>
</tr>
<tr>
<td>Alpha-d-galactofuranose</td>
<td>12.644</td>
<td>5.09</td>
<td>162</td>
<td>Furan</td>
<td></td>
</tr>
<tr>
<td>Cyclohexenone</td>
<td>12.763</td>
<td>0.56</td>
<td>152</td>
<td>Alkanone</td>
<td></td>
</tr>
<tr>
<td>Diethyl Phthalate</td>
<td>12.862</td>
<td>14.93</td>
<td>222</td>
<td>Esters</td>
<td></td>
</tr>
<tr>
<td>5-Isopropenyl-2-methylcyclopent-1-ene-carboxaldehyde</td>
<td>13.080</td>
<td>0.36</td>
<td>150</td>
<td>Aldehyde</td>
<td></td>
</tr>
<tr>
<td>7-dimethyl-octahydro-isobenzofuran-3a-ol</td>
<td>13.650</td>
<td>0.56</td>
<td>226</td>
<td>Furan</td>
<td></td>
</tr>
<tr>
<td>2-Cyclohexen-1-one</td>
<td>14.215</td>
<td>0.97</td>
<td>222</td>
<td>Alkanone</td>
<td></td>
</tr>
<tr>
<td>2-Cyclohexen-1-one</td>
<td>15.011</td>
<td>1.76</td>
<td>222</td>
<td>Alkanone</td>
<td></td>
</tr>
<tr>
<td>Cyclohexane propionic acid, 4-oxo</td>
<td>15.620</td>
<td>0.20</td>
<td>198</td>
<td>Ester</td>
<td></td>
</tr>
<tr>
<td>Compound Name</td>
<td>Molecular Formula</td>
<td>Molecular Weight (g/mol)</td>
<td>Biological Activity</td>
<td></td>
<td></td>
</tr>
<tr>
<td>------------------------------------</td>
<td>-------------------</td>
<td>--------------------------</td>
<td>-------------------------------------------------------------------------------------</td>
<td></td>
<td></td>
</tr>
<tr>
<td>7,9-Di-tert-butyl-1-oxaspiro</td>
<td></td>
<td>276</td>
<td>Oxaspiro compound</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Dibutyl phthalate</td>
<td></td>
<td>278</td>
<td>Esters</td>
<td></td>
<td></td>
</tr>
<tr>
<td>n-Hexadecanoic acid</td>
<td></td>
<td>256</td>
<td>Monocarboxylate</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Phytol;</td>
<td></td>
<td>296</td>
<td>Monocarboxylate</td>
<td></td>
<td></td>
</tr>
<tr>
<td>9,12,15-Octadecatrienoic acid</td>
<td></td>
<td>278</td>
<td>Monocarboxylate</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Octadecanoic acid;</td>
<td></td>
<td>284</td>
<td>Monocarboxylate</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Phenol;</td>
<td></td>
<td>94</td>
<td>Benzenol</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Glycerin;</td>
<td></td>
<td>92</td>
<td>Trihydric alcohol</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Table 4: Reported biological activity of compounds present in the methanol extracts of Terminalia ivorensis

<table>
<thead>
<tr>
<th>Compound Name</th>
<th>Biological Activity</th>
<th>Reference(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cyclohexanol</td>
<td>Aromatic Odour; Anti-Inflammatory Effects; Analgesic (Pain-Relieving) Effects</td>
<td>(Salakhutdinov et al., 2017).</td>
</tr>
<tr>
<td>Citronellol</td>
<td>Anti-Inflammatory Effects; Analgesic (Pain-Relieving) Effects; Antioxidant Activity; Sedative and Anxiolytic Effects;</td>
<td>(Salakhutdinov et al., 2017).</td>
</tr>
<tr>
<td>2,6-Octadien-1-ol</td>
<td>Anti-Inflammatory Effects; Sedative and Anxiolytic Effects; Anti-Infective Properties;</td>
<td></td>
</tr>
<tr>
<td>1,2,3-Benzenetriol</td>
<td>Cosmetic and Personal Care Products; Anti-Inflammatory Effects; Tyrosinase Inhibition; Acetylcholinesterase inhibition;</td>
<td>(Ozturk, 2015).</td>
</tr>
<tr>
<td>Propan-2-one;</td>
<td>Antimicrobial; anti-inflammatory activity.</td>
<td>(Zhang et al., 2015).</td>
</tr>
</tbody>
</table>
Protein content and amino acid composition of the diet affect animal growth by regulating body protein synthesis and mean body mass that primarily composes organs, red muscle, and white muscle. Proteins are needed for the build-up of body tissues and amino acids are the building blocks of proteins. In comparison to *Piper guineensis* (Benin pepper or uziza pepper), whose protein content was reported to be 8.75%, and *Morinda lucida* (Brimstone tree), whose protein content was reported to be 8.75%.

---

**D-Allose**
- Antioxidant Activity;
- Anti-Inflammatory Effects;
- Anticancer Potential;
- Diabetes Management

(Shintani et al., 2020).

**Tetradec-11-en-1-ol acetate**
- Antimicrobial Activity

(Brönner et al., 2014).

**3,3-Dimethyl-4-phenyl-4-penten-2-one**
- Antioxidant Activity;
- Anti-Inflammatory Effects

(Gambhire el al, 2010).

**Lilial**
- Potential allergen

(Alves et al, 2012)

**Benzene**
- Antimicrobial Properties;
- Antioxidant Activity;
- Anti-Inflammatory Effects;
- Analgesic (Pain-Relieving) Effects.

(Nurzynska-Wierdak et al., 2014).

**Galactofuranose**
- Antifungal Properties;
- Antigenic Determinants;
- Cell Surface Glycan

(Li et al., 2021).

**Cyclohexone**
- Insecticidal activity

(Chung et al., 2011).

**Diethyl Phthalate**
- Antimicrobial Activity

(Xuan et al., 2006).

**5-Isopropenyl-2-methylcyclopent-1-enecarboxaldehyde**
- Antimicrobial activity

(Yu et al., 2010).

**1-Isobutyl-7**
- Antimicrobial activity

(Gupta et al., 2017).

**2-Cyclohexen-1-one**
- No activity

(Alves et al., 2012).

**2-Cyclohexen-1-one**
- No activity

(Chung et al., 2011).

**Cylohexane propionic acid**
- Anti-microbial activity

(Gupta et al., 2017).

**7,9-Di-tert-buty1-1-oxaspiro**
- Irritant

(Gupta et al., 2017).

**Dibutyl phthalate;**
- Endocrine disruption

(Bhakta et al., 2009).

**n-Hexadecanoic acid**
- n-Hexadecanoic acid serves as a major energy storage molecule, a crucial metabolite for the synthesis of complex lipids, and contributes to the structure and stability of cell membranes. It also has anti-inflammatory activity.

(Muhammad et al., 2018).

**Phytol**
- Phytol shows anti-inflammatory properties, antioxidant activity, potential anti-cancer activity, and anti-microbial activity against various microorganisms. It serves as a metabolite in the human gut, produced through the action of gut bacteria, with potential health impacts.

(Parvathi et al., 2022).

**9,12,15-Octadecatrienoic acid**
- 9,12,15-Octadecatrienoic has diverse biological activities that encompass anti-inflammatory effects, cardiovascular health benefits, support for brain health, promotion of skin health, potential anti-cancer effects, and a role in weight management.

(Abbobaker and Majinda, 2016).

**Octadecanoic acid**
- Energy Storage;
- Precursor for Lipid Synthesis;
- Role in Cell Membrane Structure;
- Precursor for Lipid Synthesis;
- Role in Cell Membrane Structure.

(Keawsa-ard et al., 2012).

**Phenol**
- Antioxidant;
- Anaesthetics;
- Skin irritant.

(Álvarez-Suárez et al., 2012).

**Glycerin**
- Moisturizing agent;
- Wound healing;
- Laxatives.

(Szél et al., 2015).

The proximate analysis technique helps in understanding the fundamental characteristics and properties of a substance, which can be valuable in nutrition and energy utilization (Iloowo, et al., 2018). Protein content and amino acid composition of the diet affect animal growth by regulating body protein synthesis and mean body mass that primarily composes organs, red muscle, and white muscle. Proteins are needed for the build-up of body tissues and amino acids are the building blocks of proteins.
2.28%). *Terminalia ivorensis* leaves had a protein content of 10.94±0.03% (Table 3.1). Despite being somewhat higher than those mentioned above, the protein content of *Terminalia ivorensis* leaves is still much lower than that of *Moringa oleifera* leaves (22.99%) (Asaolu et al., 2012), *Ageratum conyzoides* leaves (18.62%), and *Anthocleista djalonensis* leaves (17.28%) (Nduche et al., 2015). This suggests that the plant *Terminalia ivorensis* is a convenient source of a protein that can be used as protein supplement. According to Khan et al. (2013), a plant's ash content is a measurement of the minerals it contains. The methanol leaf extract of *Terminalia ivorensis* has a 1.42±0.01% ash content. From this, it can be deduced that there are few or a few minerals present. According to Ahmed and Birnin-Yauri (2008), fibre in human diets is known to facilitate the flow of waste through the colon, acting as a potent anti-conspitivating agent. For the methanol leaf extract, the value for the crude fibre content from this investigation was 0.95±0.01%. The value obtained suggests that the extract is poor in fibre. Food moisture content provides important information on food preservation and storage. High moisture content aids in sustaining the life of the food composition, but it can result in microbial deterioration during storage. This explains why the majority of traditional users, including those who store *Terminalia ivorensis* in dried form (Gbadamosi et al., 2011; Ibukunoluwa et al., 2015), do so. The methanol leaf extract's moisture content was found to be 41.02±0.21% (Table 1.0). This result suggests that the moisture content is high. The methanol leaf extract's proximate analysis revealed a 7.97±0.21% lipid content. According to Ibukunoluwa et al. (2015), lipids are crucial for giving the body its maximum amount of energy, assisting in the movement of fat-soluble vitamins, insulating and protecting internal tissue, and supporting critical cellular functions. Plants contain carbohydrates in the form of nitrogen-free extract. Carbohydrates are one of the most important components in many foods. Carbohydrates may be present as isolated molecules or they may be physically associated or chemically bound to other molecules. Individual molecules can be classified according to the number of monomers that they contain as monosaccharides, oligosaccharides or polysaccharides which may be covalently linked by glycosidic bonds. Molecules in which the carbohydrates are covalently attached to proteins are known as glycoproteins, whereas those in which the carbohydrates are covalently attached to lipids are known as glycolipids. *Terminalia ivorensis* leaf extract in methanol has a high value of 37.70±0.05%. This number is considerably less than that reported by Nduche et al. (2015) for the leaves of *Anthocleista djalonensis*, which was 54.89%. However, *Ipomoea batatas* leaves were reported to contain 75% carbohydrates (Khan et al., 2013). In contrast, 94% of *Talinum triangulare* is made up of carbohydrate. The above result indicates the plant is a reliable source of energy. The mineral composition of extract as shown in Table 2.0 show the plant is rich in some minerals. Calcium plays a crucial role in the development and stability of cell walls as well as the preservation of membrane permeability and structure. It plays an important role in hormonal and nervous coordination (Felhi et al., 2016). Calcium as well as phosphorus and magnesium in the blood must be balanced, therefore the closer a food composition comes to that balance, the better for the body (Adlercreutz et al., 1993). Reduced root growth, death of terminal buds, and distorted leaves are the effects of calcium deficiency in plants (Felhi et al., 2016). Calcium is directly involved in each heartbeat and in controlling arterial pressure. The extract richness in calcium in addition to detected phytochemical (D- Allose) may have a very beneficial effect on cardiovascular health. Studies have shown that lack of calcium causes nervousness, the extract being very rich in calcium can also achieve the function of the nervous system. Magnesium, which is present in significant concentrations in the extract, functions as a cofactor for enzymes that catalyse several metabolic pathways e.g. hexokinase. Iron is a key component of heme, which functions as a prosthetic group for cytochrome and hemoglobin. Some enzymes required for metabolic activities are also activated by iron.

Fifty-seven (47) phytochemical compounds were detected by GC-MS single phase ion mode including 1, 2, 3-Benzentriol (26.31%), Dithyl Phthalate (14.93%), D-Allose (6.96%), Anhydro-alpha-d-galactofuranose (5.09), and Glyceralin (6.28%) as the most predominant. According to Li et al. (2002) dithyl phthalate exhibits antibacterial and antioxidant activity. According to Ozturk (2015), 1, 2, 3-Benzentriol have anti-inflammatory and antioxidant properties as well as the ability to inhibit tyrosinase and acetylcholinesterase. D-Allose (C₄H₂O₅) (6.96%) exhibits antioxidant, anti-inflammatory and anti-cancer activities, with potential for managing diabetes (Shintani et al., 2020; Khadmim, 2016). N-hexanedecanoic acid (1.38%), often referred to as palmitic acid, is a key energy storage molecule that is also essential for the creation of complex lipids and helps keep cell membranes stable. It also has anti-inflammatory properties (Aparna et al., 2012; Biermann et al., 2000). Phytol demonstrates anti-inflammatory, antioxidant, possible anti-cancer, and antimicrobial activities against a variety of bacteria. It functions as a metabolite in the human gut and is produced by gut bacteria, which may have its own impact in gut health (Muhammad et al., 2018). According to Nurmeyna-Wierta et al. (2014), 1, 6-Anhydro- alpha-d-galactofuranose (C₄H₆O₅) (5.09%) functions as a cell surface glycans, has anti-fungal characteristics, and is employed as an antigenic determinant. Octadecanoic acid (0.37%) functions as an energy storage molecule, a molecule that enhances the structure of cell membranes, and a precursor for lipids synthesis, and antiviral activity and it has a role in cellular signalling (Keawsa-ard et al., 2012). Medicinal plants are known to play a very important role in diseased conditions by exhibiting anti-oxidative properties. The results obtained from this study is in agreement with the report by Enadeghe and Omoregie (2011) that *Terminalia catappa* possesses bioactive with antioxidant, antiviral, antimicrobial and anti-inflammatory activities. The antioxidant activity could be related to the presence of hydroxyl groups, double bonds conjugation and resonance effect in the structure of the phenolic compounds detected in the extract. *Terminalia ivorensis* has also been shown to have antioxidant and anti-inflammatory potential as reported by Aparna et al., 2012; Biermann et al., 2000). The antiviral activity of octadecanoic acid has been shown to inhibit the enzyme required for viral replication and cells invasion as reported by Keawsa-ard et al., 2012. These detected compounds have OH groups and oxygen atoms C=O with the ability to donate hydrogen atoms to free radicals, neutralize their scavenging effect, and chelate metals. The presence of alternating single and multiple bonds in detected structures as shown in (table 3 and figure 2- 6) provides resonance stability to the molecule, thus allowing easy interaction with reactive oxygen species (ROS) with a powerful antioxidant potential. Phenolic compounds also have vasorelaxation and anti-allergenic activity (D’Archivio et al., 2010).
CONCLUSION
The study provides scientific evidence that the methanol leaf extract of Terminalia ivorensis contains various volatile phytochemical compounds with reported medicinal potential. The structure, functional groups and resonance effect of these detected bioactive compounds may be responsible for the various reported antioxidants, biological and therapeutic potential of the plant extract. Thus may be considered as a potential drug, feed or fertilizer target.

ACKNOWLEDGEMENT
The authors acknowledge the laboratory staff of Department of Biochemistry University of Benin, for their support during the laboratory work.

REFERENCES


Doughari, J.H. (2012). Phytochemicals: extraction methods, basic structures and mode of action as potential chemotherapeutic agents (pp. 1-33). Rijeka, Croatia: INTECH Open Access Publisher.


